

Stereoselective Synthesis of Peptidyl Trifluoromethyl Alcohols and Ketones: Inhibitory Potency Against Human Leucocyte Elastase, Cathepsin G, Porcine Pancreatic Elastase and HIV-1 Protease

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Abstract

New fluorinated inhibitors have been designed to target two major proteases—human leucocyte elastase and HIV-1 protease.

Two series of β -peptidyl trifluoromethyl alcohols (TFMAs) Z-L-Val-NH-*CH(Y)-*CH(OH)-CF₃, where *C is the chiral centre, varied in the nature of the substituent Y, a phenylethyl [-(CH₂)₂-C₆H₅] or an isopropyl [-CH(CH₃)₂] group. These TFMAs were first synthesized as two pairs of the *syn* and *anti* diastereoisomers. The inhibitory effects of these mixtures were then assessed on three serine proteases chosen on the basis of the aromatic and aliphatic nature of the substituents—human leucocyte elastase (HLE), human cathepsin G (HCG) and porcine pancreatic elastase (PPE). In the presence of detectable inhibition, each epimer at C2 was separated to determine its inhibition constant (K_i) towards HLE, HCG and PPE. The stereoisomerically pure TFMAs were then oxidized into peptidyl trifluoromethyl ketones (TFMKs) for similar inhibition assays. The absolute configuration of the compounds remained unknown. One epimer at C2 of each *syn* and *anti* TFMA with the phenylethyl substituent behaved as a competitive inhibitor towards HLE and HCG with inhibition constants below the millimolar range, whereas their TFMK counterparts were non-inhibitors. In the second series, the two ketones inhibited both elastases with K_i values in the micromolar range, whereas only the *syn* TFMA was active towards HLE (K_i = 5.65 × 10⁻⁴ M). The tested compounds also had structural properties compatible with recognition by HIV-1 protease. The inhibition of the enzyme was observed with TFMK only (IC₅₀ = 15–200 μM). The phenylethyl substituent promoted inhibition by a factor of 10 (IC₅₀ = 15 μM) compared with the isopropyl substituent (IC₅₀ = 200 μM) leading to selective inhibition of HIV-1 protease.

Isomerically pure TFMKs were more potent towards HLE than the alcohols from which they were obtained. However, an enantiomerically pure TFMA selectively inhibited HLE unlike its TFMK analogue which also inhibited PPE. This last result together with the selective inhibition of HIV-1 protease by TFMK with a phenylethyl substituent might be relevant to the design of specific HLE and HIV-1 inhibitors as therapeutic agents.

In this study our three target serine proteases were human leucocyte elastase (HLE, EC 3.4.21.37),

pancreatic porcine elastase (PPE, EC 3.4.21.36) and human leucocyte cathepsin G (HCG, EC 3.4.21.20). PPE and HLE share 43% sequence homology and are structurally very similar (Bode et al 1989). HLE and HCG originate from the azurophilic granules of polymorphonuclear neutrophils (Dewald et al 1975). HLE preferentially cleaves

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after aliphatic amino acid residues of medium size such as valine, PPE after aliphatic amino acids of smaller size such as alanine or norvaline; HCG has chymotrypsin-like specificity, favouring bulkier aliphatic or aromatic residues, such as phenylalanine, tryptophan or leucine, in this position (Powers et al 1977).

During inflammation, elastin and other essential components of the connective tissue are degraded by HLE released in abundance by polymorphonuclear neutrophils (Bieth 1989). HCG is thought to act in synergy with HLE in these degradative processes (Boudier et al 1981). The breakdown of the extracellular matrix by HLE is usually prevented by α_1 -proteinase inhibitor, the primary physiological plasmatic inhibitor of this elastase (Travis & Salvesen 1983). However, genetic or functional deficiencies of α_1 -proteinase inhibitor can occur and are considered to be responsible for inflammatory diseases such as pulmonary emphysema and chronic bronchial injury (Snider 1987).

In recent years, considerable efforts have been made to design synthetic inhibitors of HLE as drugs for elastase-induced diseases (Edwards & Bernstein 1994). The design of several peptidic inhibitors has been based on the structural requirements of the binding sites of HLE for its peptidic substrates (Bode et al 1989). HLE inhibitors include, among others, irreversible inhibitors such as peptidyl chloromethylketones (Powers et al 1977) and mechanism-based inhibitors such as monocyclic β -lactams (Maillard et al 1990; Doherty et al 1993). Efficient reversible inhibitors have also been reported, for example trifluoroacetylpeptide anilides (Mattos et al 1995; Amour et al 1996).

Transition-state analogues such as peptidyl trifluoromethyl ketones (TFMKs, Imperiali & Abeles 1986; Stein et al 1987) are a powerful class of inhibitors because they match the tetrahedral intermediate occurring during proteolytic hydrolysis of a peptide substrate and thus fit efficiently into the active site of the enzyme (Krantz 1992). Because it is the serine protease that transforms TFMKs into stable tetrahedral structures, they are considered to be irreversible inhibitors (Krantz 1992). They inhibit HLE by combining with the active-site serine (Ser-195) to form a hemiketal that resembles the transition state resulting from acylation of serine by peptide substrates. The mechanism proposed involves initial binding of the unhydrated TFMK to the enzyme followed by the reversible formation of the hemiketal by addition of the O_γ of Ser-195 (Brady & Abeles 1990). The X-ray crystal structures of the complexes of PPE with Ac-L-Ala-L-Pro-L-Val-CF₃ (Takahashi et al 1988), and of chymotrypsin with Ac-D,L-Phe-CF₃ or with

Ac-L-Leu-D,L-Phe-CF₃ (Brady et al 1990) confirmed the formation of the tetrahedral ketal carbon of the inhibitors. A new method for the stereoselective synthesis of peptidyl trifluoromethyl ketones enabled the preparation of the two TFMKs of absolute stereochemistry Z-L-Val-L-Pro-L-Val-CF₃ and Z-L-Val-L-Pro-D-Val-CF₃ (Edwards 1992). Their respective inhibition activities were evaluated towards HLE. The isomer with the natural L configuration had a K_i of 0.8 nM, which was two hundred times more potent than the D isomer. This study thus revealed the importance of the chirality of the carbon C β adjacent to the ketone on the affinity with HLE. In contrast, only racemic mixtures of related β -peptidyl trifluoromethyl alcohols (TFMAs) have been evaluated towards chymotrypsin (Imperiali & Abeles 1986) and towards HLE (Stein et al 1987). Although they were found to be less active than their corresponding TFMKs, the influence of their stereochemistry on inhibitory potency remained unknown. Some difluoroketone peptides have been described as very potent inhibitors of HIV-1 protease (Sham et al 1993). This protease, activity of which is a prerequisite for viral replication, is a major target against AIDS. Because of their low molecular weights it is hoped that the biodisposition of TFMA and TFMKs will be better than that of difluoroketone peptides.

In this study, we investigate the inhibition properties towards serine proteases and HIV-1 protease of enantiomerically pure TFMA and compared their activity with those of the corresponding TFMKs. Two different series of TFMA were first synthesized as mixtures of *syn* or *anti* enantiomers. The inhibition activities of these mixtures were then checked on HLE, HCG and PPE. When inhibition was detected, the enantiomers were separated, thus enabling the determination of the inhibition constants (K_i) of the stereoisomerically pure TFMA. All four TFMKs were assayed as isomerically pure forms. The aim of this study was to compare the influence of the chirality of these compounds on their inhibition of the three serine proteases chosen. However, the absolute stereochemistry of all the compounds remained undefined.

In the first series, the four TFMA stereoisomers of Z-L-Val-NH-C*H(CH₂-CH₂-C₆H₅)-C*H(OH)-CF₃ were obtained—the *syn* pair **1a** and **1b** and the *anti* pair **2a** and **2b**. The corresponding ketones were **3a** and **3b**. In the second series, the ethylphenyl group was replaced by an isopropyl group. The TFMA obtained were the *syn* pair **4a** and **4b** and the mixture of the *anti* stereoisomers **5**, epimers at C2. The two TFMK isomers **6a** and **6b** were also prepared.

Materials and Methods

Enzyme-inhibition procedures

The enzymes HLE, PPE and HCG were purchased from Elastin Products (Owensville, MO), Serva (Fontenay-sous-Bois, France) and Novabiochem (Meudon, France), respectively. Their active site concentrations were 20, 16 and 25 nM, respectively. Their chromogenic substrates (Sigma, Saint-Quentin-Fallavier, France) were methoxy-succinyl-alanyl-alanyl-prolyl-valyl-*p*-nitroanilide (CH₃O-Suc-Ala-Ala-Pro-Val-*p*-NA) used at concentrations ranging from 0.05 to 1 mM for HLE, *N*-succinyl-alanyl-alanyl-alanyl-*p*-nitroanilide (0.05 to 1 mM) for PPE and *N*-succinyl-alanyl-alanyl-prolyl-phenylalanyl-*p*-nitroanilide (5 to 50 μM) for HCG. The enzymes were assayed at pH 8.0 and 25 °C in 1 mL buffer—0.1 M HEPES, 0.5 M NaCl, 0.1% (v/v) Tween 80, 2% (v/v) DMSO for HLE and HCG and 0.1 M Tris, 4% (v/v) DMSO for PPE. The amidolytic activities of the three proteases towards their respective chromogenic substrates were followed by use of a Lambda 5 Perkin-Elmer UV-vis spectrophotometer equipped with a thermostatted holder. The initial velocity, *v*, was determined by continuous monitoring of the release of *p*-nitroaniline at 405 nm in the presence of different concentrations of inhibitors: 0–120 μM for **1a**, 0–60 μM for **1b**, 0–300 μM for **2a**, 0–400 μM for **2b**, 0–300 μM for **3a**, 0–500 μM for **3b**, 0–300 μM for **4a**, 0–450 μM for **4b**, 0–200 μM for **5**, 0–100 μM for **6a** and 0–200 μM for **6b**. Control experiments without inhibitor were performed under the same conditions.

HIV-1 protease was kindly supplied by Rhône Poulenc Rorer (Vitry-sur-Seine, France). Kinetic assays were performed with the chromogenic substrate VSQNF(NO₂)PIV as described by Quasmi et al (1997). To determine IC₅₀ (the concentration resulting in 50% inhibition), percentage inhibition was measured with different concentrations of inhibitors—1.25 to 12.5 μM **3a** and **3b** and 62.5–250 μM **6a** and **6b**.

Kinetic analysis of enzyme inhibition

The double-reciprocal plots of *v* against substrate concentration were straight lines with *x* intercept of $-1/K'_M$. The apparent Michaelis constants K'_M , equal to $K_M \times (1 + [I]_0/K_i)$, and the maximum velocity V_M were estimated by iterative least-squares fitting to the equation for competitive inhibition $v = V_M \times [S]_0 / ([S]_0 + K'_M)$ using the software package KaleidaGraph 2.1.3 (Abelbeck Software, Reading, PA). The constant K_i and its standard deviation were determined from the linear plot of K'_M/V_M against inhibitor concentration

(least-squares analysis). Initial estimates of K_i were calculated from Dixon plots. Determination of IC₅₀ values was performed using at least four concentrations of inhibitor, and by fitting the data to the equation:

$$\% \text{ inhibition} = (100 \times [I]_0) / (IC_{50} + [I]_0).$$

Chemical synthesis

Procedure. Melting points were recorded on a Köfler apparatus. ¹⁹F NMR spectra were obtained at 84.6 MHz and ¹⁹F chemical shifts are reported in ppm, negative up-field relative to internal CFCl₃. The ¹H NMR and ¹³C NMR measurements were performed on Brüker AC200 and ARX400 instruments. The chemical shifts are reported in ppm, either downfield from tetramethylsilane (TMS) internal standard, or from the attributed chemical shifts of the deuterated solvents: CDCl₃ (7.26 ppm) and CD₃OD (3.50 ppm), as indicated. The coupling constants are given in Hz. Analytical thin-layer chromatography (TLC) was performed on aluminium plates coated with silica gel containing a fluorescent indicator (no. 5554, Merck; 254 nm). For all compounds visualization of the spots after TLC runs was performed by illumination with UV light (254 nm) or by spraying with ninhydrin and heating.

Because the Dess-Martin periodinane 1,1,1-triacetoxy-1,1-dihydro-1,2-benziodoxol-3(1*H*)-one is no longer commercially available it was prepared in our laboratory by the modified procedure reported elsewhere (Ireland & Liu 1993).

The *syn* and *anti* trifluoromethyl amino alcohols **7a** and **7b**, required for the peptidic coupling reaction, were prepared according to the general procedure reported by our laboratory (Bégué et al 1992).

syn and *anti* *N*-Benzyloxycarbonyl-*N*-(3,3,3-trifluoro-2-hydroxy-1-(isopropyl)-propyl) *L*-valinamides **4** and **5**. A solution of trifluoromethyl amino alcohol **7a** or **7b** (0.5 g; 2.9 mmol), CBZ-*L*-valine (0.7 g; 2.8 mmol) and 1-hydroxybenzotriazole (HOBT; 0.4 g; 2.9 mmol) in CH₂Cl₂ (5 mL) was cooled to 0 °C and dicyclohexylcarbodiimide (DCC; 0.6 g; 2.9 mmol) was added. The reaction mixture was stirred at 0 °C for 1 h and at room temperature overnight. After extraction with AcOEt, the organic layer was washed successively with 10% aqueous HCl, a saturated solution of NaHCO₃ and brine, and dried over MgSO₄. The residue was then chromatographed on silica gel (CH₂Cl₂-AcOEt, 85 : 15) to give alcohol **4** or **5**.

The mixture of the two diastereoisomers *syn* β -peptidyl alcohols **4a** and **4b** (0.9 g; 75%) were obtained as a white powder. The diastereoisomers were separated by re-crystallization from CH_2Cl_2 -cyclohexane (1:5); mp **4a** 116–118 °C, **4b** 100–102 °C.

The mixture of the two diastereoisomers *anti* β -peptidyl alcohols **5** (0.6 g; 51%) were obtained as a white powder and were not separated; mp 150–152 °C.

syn β -Peptidyl alcohols **4a** and **4b**

Diastereomer 4a (RS or SR). ^1H NMR (CDCl_3 ; TMS): 0.95, d (6.8), 12H (4 CH_3); 1.67, m, 2H (2NH); 2, m, 2H ($\text{CH}(\text{CH}_3)_2$); 3.8–4.1, m, 2H (CHNH); 5, AB system (12.1), 2H (OCH_2Ph); 5.45, m, 1H (CHCF_3); 7.2, m, 5H (H arom.). ^{13}C NMR (CDCl_3): 17.8; 17.9; 18.7; 18.9; 19.1; 29.9; 53.6; 61.1; 67.2; 68.5, q ($^2\text{JCF} = 29$ Hz), CHCF_3 ; 125, q ($^1\text{JCF} = 286$ Hz), CF_3 ; 127.8; 128.1; 128.4; 135.9; 156.8; 171.8. ^{19}F NMR (CDCl_3 , CFCl_3): –77.1, d ($^3\text{JHF} = 6.9$ Hz). Analysis: calc. for $\text{C}_{19}\text{H}_{26}\text{F}_3\text{N}_2\text{O}_4$ C, 56.57%; H, 6.50%; N, 6.94%; found C, 56.60%; H, 6.54%; N, 7.08%.

Diastereoisomer 4b (SR or RS). ^1H NMR (CDCl_3 ; TMS): 0.95, m, 12H (4 CH_3); 2, m, 2H (2NH); 2.2, m, 2H ($\text{CH}(\text{CH}_3)_2$); 4–4.2, m, 2H (CHNH); 5.15, AB system (12.4), 2H (OCH_2Ph); 5.5, m, 1H (CHCF_3); 7.2, m, 5H (H arom.). ^{13}C NMR (CDCl_3): 18; 18.2; 18.7; 18.8; 20.2; 31; 54; 60; 68; 68.7, q ($^2\text{JCF} = 30$ Hz), CHCF_3 ; 124, q ($^1\text{JCF} = 286$ Hz, CF_3); 127.8; 128.2; 128.4; 136; 157; 172. ^{19}F NMR (CDCl_3 , CFCl_3): –77.2, d ($^3\text{JHF} = 7$ Hz). Analysis: calc. for $\text{C}_{19}\text{H}_{26}\text{F}_3\text{N}_2\text{O}_4$ C, 56.57%; H, 6.50%; N, 6.94%; found C, 56.62%; H, 6.52%; N, 7.05%.

Anti β -peptidyl alcohols 5 (SS and RR). ^1H NMR (CDCl_3 ; TMS): 0.9, m, 12H (4 CH_3); 1.65, m, 2H (2NH); 1.9–2.1, m, 2H ($\text{CH}(\text{CH}_3)_2$); 4, m, 2H (CHNH); 5, s, 2H (OCH_2Ph); 5.25, m, 1H (CHCF_3); 7.25, m, 5H (H arom.). ^{13}C NMR (CDCl_3): 19.6; 19.9; 20.2; 20.3; 29.4; 30.8; 57.7; 62.1; 68.3; 70, q ($^2\text{JCF} = 30$ Hz), CHCF_3 ; 124.6, q ($^1\text{JCF} = 290$ Hz), CF_3 ; 128.4; 129; 136; 136.8; 157.6; 173.9. ^{19}F NMR (CDCl_3 , CFCl_3): –74.3, d ($^3\text{JHF} = 7$ Hz).

Oxidation of β -peptidyl trifluoromethyl alcohols 1 and 4 to their corresponding ketones 3 and 6. A solution of alcohol **1** or **4** (0.4 mmol) in CH_2Cl_2 (5 mL) was added to a stirred suspension of Dess-Martin periodinane (4 equiv) in CH_2Cl_2 (5 mL) and stirred at room temperature overnight. The reaction mixture was diluted with AcOEt and poured into

saturated aqueous NaHCO_3 (20 mL) containing a sevenfold excess of $\text{Na}_2\text{S}_2\text{O}_3$ (15 mL). The organic layer was washed with brine and dried over MgSO_4 . The residue was chromatographed on silica gel (CH_2Cl_2 -*i*-PrOH, 95:5). The TFMA **1a** (0.2 g) was treated with Dess-Martin reagent (0.680 g) to give, after chromatography on SiO_2 , the ketone **3a** (0.127 g; 93%) as a white powder, mp 146–148 °C. The TFMA **1b** (0.2 g) was treated with Dess-Martin reagent (0.680 g) to give, after chromatography on SiO_2 , the ketone **3b** (0.120 g; 85%) as a white powder, mp 130–132 °C. The TFMA **4a** (0.160 g) was treated with Dess-Martin reagent (0.680 g) to give, after chromatography on SiO_2 , the ketone **6a** (0.130 g; 80%) as a white powder, mp 100–102 °C. The TFMA **4b** (0.160 g) was treated with Dess-Martin reagent (0.680 g) to give, after chromatography on SiO_2 , the ketone **6b** (0.140 g; 90%) as a white powder, mp 108–110 °C.

N-Benzyloxycarbonyl-N-(3,3,3-trifluoro-2-hydroxy-2-oxo-1-(phenylethyl)-propyl) L-valinamides 3a and 3b. **3a** (S or R). ^1H NMR (CD_3OD ; TMS): 0.9, d (6.7), 6H (2 CH_3); 1.7–1.95, m, 3H (BnCH_2 and $\text{CH}(\text{CH}_3)_2$); 2.5, m, 2H (PhCH_2); 3.9, m, 2H (CHCOCF_3 and CH-val); 5, AB system (12), 2H (OCH_2Ph); 7.2, m, 10H (H arom.). ^{19}F NMR (CD_3OD , CFCl_3): –76.2, s. Analysis: calc. for $\text{C}_{24}\text{H}_{27}\text{F}_3\text{N}_2\text{O}_4$ C, 62.06%; H, 5.86%; N, 6.03%; found C, 62.17%; H, 5.88%; N, 6.16%.

3b (R or S). ^1H NMR (CD_3OD ; TMS): 0.9, d (6.8), 6H (2 CH_3); 1.8–2.1, m, 3H (BnCH_2 and $\text{CH}(\text{CH}_3)_2$); 2.6, m, 2H (PhCH_2); 3.8, m, 2H (CHCOCF_3 and CH-val); 5.2, AB system (12), 2H (OCH_2Ph); 7.4, m, 10H (H arom.). ^{19}F NMR (CD_3OD , CFCl_3): –76.1, s.

N-Benzyloxycarbonyl-N-(3,3,3-trifluoro-2-hydroxy-2-oxo-1-(isopropyl)-propyl) L-valinamides 6a and 6b. **6a** (S or R). ^1H NMR (CDCl_3 ; TMS): 0.9, m, 12H (2(CH_3) $_2$); 1.2–1.6, m, 2H (2NH); 2.1–2.3, m, 2H (2 $\text{CH}(\text{CH}_3)_2$); 4.9, m, 1H (CHCOCF_3); 5.2, m, 1H (CH-*i*Pr); 5.1, s, 2H (OCH_2Ph); 7.3, m, 5H (H arom.). ^{13}C NMR (CDCl_3): 17.6; 17.9; 19.1; 19.6; 29.3; 30.7; 58.7; 60.4; 67.2; 120.0, q ($^1\text{JCF} = 285$ Hz), CF_3 ; 128; 128.2; 128.6; 136.1; 156.6; 171.88; 190.0, d ($^2\text{JCF} = 30$ Hz), COCF_3 . ^{19}F NMR (CDCl_3 , CFCl_3): –76.6, s. Analysis: calc. for $\text{C}_{19}\text{H}_{24}\text{F}_3\text{N}_2\text{O}_4$ C, 56.85%; H, 6.03%; N, 7.0%; found C, 56.89%; H, 6.07%; N, 7.06%.

6b (R or S). ^1H NMR (CDCl_3 ; TMS): 0.9–1.2, m, 12H (2(CH_3) $_2$); 2.2–2.4, m, 2H (2 $\text{CH}(\text{CH}_3)_2$); 4.1, m, 1H (CH-*i*Pr); 4.8, dd (6.2; 2), 1H (CHCOCF_3); 5.1, s, 2H (OCH_2Ph); 5.3, broad s, 2H (NH); 7.3, m,

5H (H arom.). ^{13}C NMR (CDCl_3): 17.7; 17.8; 19.4; 19.9; 29.3; 30.5; 58.8; 60.0; 67.5; 118.8, q ($^1\text{JCF} = 285$ Hz), CF_3 ; 128; 128.2; 128.4; 128.7; 136.2; 156.7; 171.9; 192.0, d ($^2\text{JCF} = 28$ Hz), COCF_3 . ^{19}F NMR (CDCl_3 , CFCl_3): -76.5, s. Analysis: calc. for $\text{C}_{19}\text{H}_{24}\text{F}_3\text{N}_2\text{O}_4$ C, 56.85%; H, 6.03%; N, 7.0%; found C, 56.87%; H, 6.05%; N, 7.05%.

Results and Discussion

Chemistry

The synthesis of the homochiral *syn* and *anti* β -peptidyl TFMA 1 and 2 (Figure 1) has previously been reported (Bégué et al 1994). The *N*-CBZ-protected dipeptide analogues **1a** and **1b** were oxidized to the corresponding α -peptidyl TFMKs **3a** and **3b** (Figure 2) by Dess-Martin periodinane (Ireland & Liu 1993) under Linderman conditions (Linderman & Graves 1989). It is important to underline that with the Dess-Martin reagent no epimerization at the chiral center α to the ketone occurred, according to ^{19}F NMR.

The *syn* and *anti* β -peptidyl TFMA 4 and 5 were conveniently prepared as illustrated in Figure 3. Condensation of the *N*-CBZ-valine with the *syn*

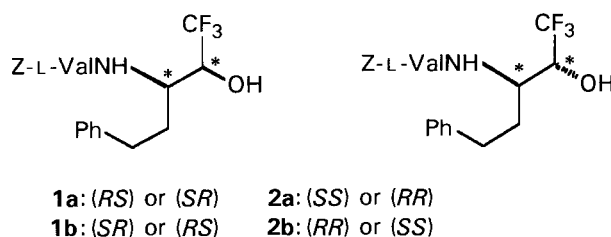


Figure 1. Structures of the *syn* and *anti* β -peptidyl trifluoromethyl alcohols **1** and **2**.

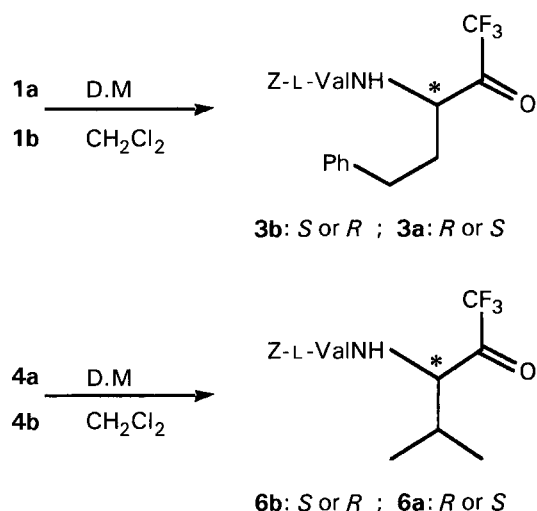


Figure 2. Preparation of the α -peptidyl trifluoromethyl ketones **3a**, **3b**, **6a** and **6b**. D.M = Dess-Martin reagent.

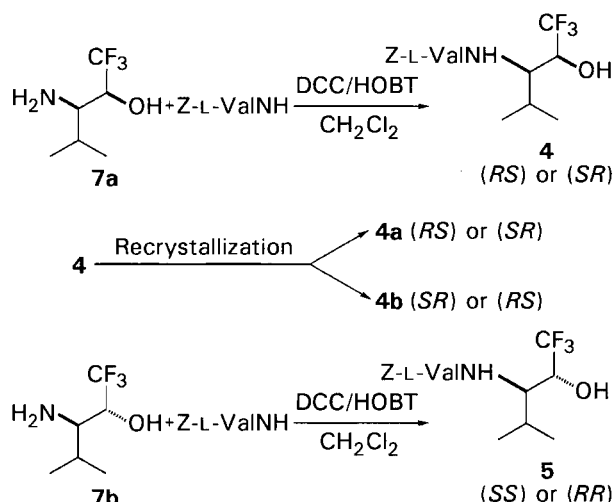


Figure 3. Preparation of the *syn* and *anti* β -peptidyl trifluoromethyl alcohols **4** and **5**.

trifluoromethyl amino alcohol **7a** in the presence of dicyclohexylcarbodiimide and 1-hydroxybenzotriazol (DCC-HOBT) in CH_2Cl_2 gave the *syn* peptidyl alcohols **4**, which were then separated by re-crystallization from a 1:5 mixture of CH_2Cl_2 and cyclohexane to give the homochiral alcohols **4a** and **4b**. In a similar manner the *anti* β -peptidyl TFMA 5 were efficiently obtained as a mixture of stereoisomers from the *anti* trifluoromethyl amino alcohol **7b**. Oxidation of the homochiral alcohols **4a** and **4b** with Dess-Martin periodinane gave the corresponding α -peptidyl TFMKs **6a** and **6b** (Figure 2).

Biochemical studies

In the first series, TFMA **1a** and **2a** competitively inhibited HLE and HCG (K_i between 0.13 and 0.94 mM) but not PPE whereas their respective enantiomers did not inhibit any of the enzymes. HLE and HCG have different structural requirements, yet the two enzymes had similar affinities towards **1a** and **2a** (Table 1). For both enzymes, **1a** had the highest potency. In comparison, the racemic mixture of TFMA Ac-Leu-*R,S*-CH($\text{CH}_2\text{C}_6\text{H}_5$)-*R,S*-CH(OH)- CF_3 also behaved as a reversible competitive inhibitor with a K_i of 0.2 mM towards chymotrypsin, whereas its ketone analogue Ac-Leu-*D,L*-Phe- CF_3 was a slow-binding inhibitor ($K_i = 1.2$ μM , Imperiali & Abeles 1986). However, neither of the ketone analogues of **1a** and **2a** inhibited chymotrypsin.

The second series of compounds containing an isopropyl group was designed to target HLE. Among the TFMA, only **4a** competitively inhibited HLE (Figure 4); it had no inhibitory activity towards PPE or HCG. In contrast, TFMKs **6a** and

Table 1. K_i values (M) for inhibition of serine proteases and HIV-1 protease by trifluoromethyl alcohols and ketones at 25 °C and pH 8.0.

Configuration	Compound	Human leucocyte elastase	Porcine pancreatic elastase	Human cathepsin G	HIV-1 protease†
<i>syn</i> R^*,S^*	1a	1.26×10^{-4}	—	1.98×10^{-4}	—
<i>syn</i> S^*,R^*	1b	—	—	—	—
<i>anti</i> S^*,S^*	2a	9.45×10^{-4}	—	6.08×10^{-4}	—
<i>anti</i> R^*,R^*	2b	—	—	—	—
S^*	3a	—	—	—	1.5×10^{-5}
R^*	3b	—	—	—	1.5×10^{-5}
<i>syn</i> R^*,S^*	4a	5.65×10^{-4}	—	—	—
<i>syn</i> S^*,R^*	4b	—	—	—	—
<i>anti</i> S,S and R,R	5	—	—	—	—
S^*	6a	2.37×10^{-6}	4.39×10^{-5}	—	2×10^{-4}
R^*	6b	8.30×10^{-6}	7.69×10^{-5}	—	2×10^{-4}

The s.d. was < 25%. — = No inhibition. † IC₅₀ (M) determined at 30 °C and pH 4.5.

6b inhibited HLE and PPE as reversible competitive inhibitors with affinities two orders of magnitude higher than that of **4a**. Inhibitor **6a** was more potent than **6b**, by factors of 3.5 and 1.8 for HLE and PPE, respectively (Figure 5). The chirality of the $C\beta$ carbon of **6a** and **6b** thus had less effect on the inhibition of HLE than previously observed with the slightly longer TFMK Z-L-Val-L-Pro-L-Val-CF₃ and its epimer Z-L-Val-L-Pro-D-Val-CF₃ (Edwards 1992).

Both TFMA and TFMKs can be considered as transition-state analogues that mimic the tetrahedral intermediate formed before the acylation and deacylation of the serine protease by a peptide substrate. However, TFMKs are covalently bound to the serine protease during the formation of the tetrahedral hemiketal with Ser-195 and are thus irreversible inhibitors (Krantz 1992), unlike TFMA which are reversible inhibitors. Hence, TFMKs and TFMA are members of two different classes of enzyme inhibitors. It was thus interesting to compare their respective inhibitory powers as isomerically pure forms. Our study was consistent with previous studies performed with racemic mixtures showing that TFMKs are overall much more potent as serine protease inhibitors than the TFMA from which they are derived. To the best of our knowledge, the racemic mixture of a TFMK (ICI-200 880) is the only HLE-inhibitor to have progressed into clinical trials for the treatment and prevention of pulmonary emphysema (Edwards & Bernstein 1994). Our study also showed that in its effect on affinity and specificity towards HLE stereochemistry was more important for the TFMA than for the related TFMKs.

Several studies of hydrolytic enzymes (acetylcholine esterase, carboxypeptidase, angiotensin convertase or pepsin) suggest that molecules containing a di- or trifluoromethyl ketone group are better inhibitors than similar compounds lacking

fluorine (Gelb et al 1985). For aspartic proteases TFMA and TFMK might be considered as a structural mimics of the tetrahedral transition state of the hydrolytic process. The presence of a fluor-

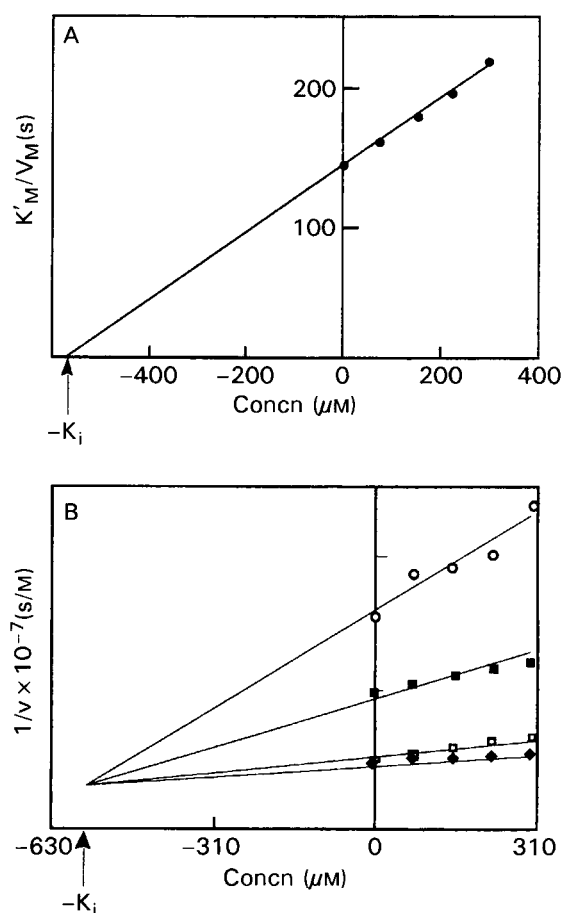


Figure 4. Inhibition of HLE by β -peptidyl trifluoromethyl alcohol **4a** at pH 8.0 and 25 °C. A. Plots of K_M^{app}/V_M against inhibitor concentration. The values of K_M^{app}/V_M were obtained from the slopes of the linear double-reciprocal plots of v against substrate concentration. B. Dixon plot of the inhibition of HLE by **4a** with CH₃O-Suc-Ala-Ala-Pro-Val-p-NA as substrate. Substrate concentrations (mM) were: ● 0.05; ○ 0.1; ■ 0.2; □ 0.5; ◆ 1.0.

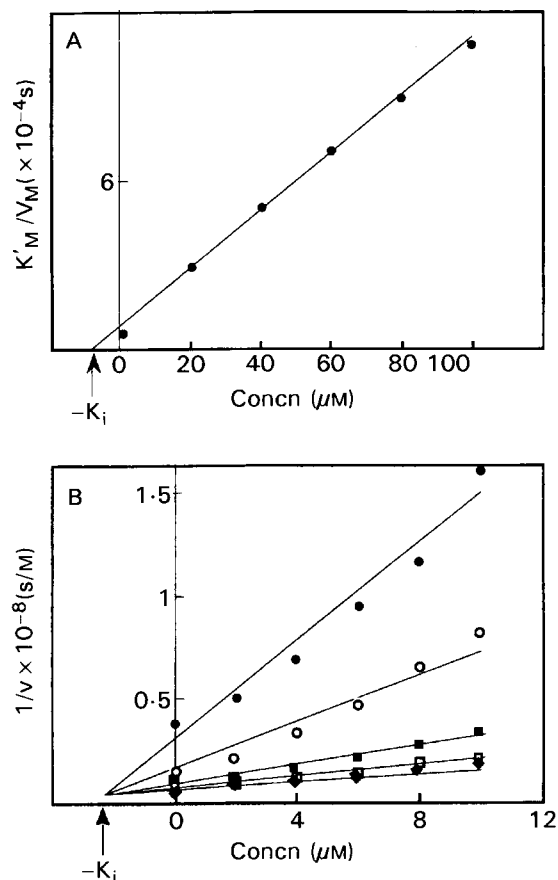


Figure 5. Inhibition of HLE by β -peptidyl trifluoromethyl ketones **6a** and **6b** at pH 8.0 and 25 °C. A. Plots of K_M^{app}/V_M against inhibitor concentration (**6b**). The values of K_M^{app}/V_M were obtained from the slopes of the linear double-reciprocal plots of v against substrate concentration. B. Dixon plot of the inhibition of HLE by **6a** with $\text{CH}_3\text{O-Suc-Ala-Ala-Pro-Val-p-NA}$ as substrate. Substrate concentrations (mM) were: \bullet 0.1; \circ 0.2; \blacksquare 0.3; \blacklozenge 0.4.

ine atom on the carbon adjacent to the carbonyl group induces an increase in the electrophilicity of the ketone group. Consequently, di- and trifluoromethyl ketones are mostly in a hydrated form in aqueous medium. TFMAs corresponding to TFMKs did not inhibit the protease within solubility limits. Inhibition of HIV-1 protease was observed only with TFMK compounds ($\text{IC}_{50} = 15\text{--}200\ \mu\text{M}$). The phenylethyl substituent resulted in tenfold greater activity ($\text{IC}_{50}\ 15\ \mu\text{M}$) than did the isopropyl substituent ($\text{IC}_{50}\ 200\ \mu\text{M}$), whatever the chirality of the compound tested. Interestingly, TFMKs bearing a phenylethyl substituent selectively inhibited HIV-1 protease with no inhibition of HLE, PPE and HCG. This might be related to the structural requirements of HIV-1 protease to accommodate substrates and inhibitors in its active site (hydrophobic groups at the P1/P'1 position).

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